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POST GRADUATE DIPLOMA IN ANALYTICAL CHEMISTRY (PGDAC)

Term-End Examination June. 2023

MCH-002: SEPARATION METHODS

Time: 3 Hours Maximum Marks: 75

Note:(i) Attempt any **five** questions.

- (ii) Marks allotted to parts are indicated on R.H.S.
- 1. (a) Define the process of separation. List any six properties used for separation.5
 - (b) Explain classification of separation methods based on rate processes giving suitable examples of each type.
 - (c) Under what conditions distribution co-efficient, thermodynamic distribution coefficient and distribution ratio, in a solvent extraction process are the same? 5

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role in extraction.

(c)

What is a diluent? Give its one example.

What are the uses of diluents?

4.	(a)	Briefly explain the synthesis of cation
		exchangers giving suitable examples. 5
	(b)	Briefly describe the technique of paper
		chromatography. Give the expression for
		the retardation factor. 5
	(c)	Discuss the applications of paper
		chromatography. 5
5.	(a)	How can you separate a mixture of protein,
		sucrose and calcium chloride from the
		aqueous solution to its components using a
		combination of membrane processes? 5
	(b)	Compare the techniques of gas
		chromatography and HPLC. 5
	(c)	Give any <i>five</i> advantages of HPLC. 5
6.	(a)	Discuss the working of a flame ionisation
		detector. Give its <i>two</i> applications. 5
	(b)	Briefly describe different types of
		adsorbents used in liquid-solid
		chromatography. 5
	(c)	Explain the technique of frontal analysis.

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7.	(a)	What are the requirements of a suitable
		internal standard in the quantitative
		analysis using internal standardisation
		method? 5
	(b)	Discuss Donnan effect. 5
	(c)	Briefly explain the important properties of
		gels. 5
8.	(a)	Discuss the unique features of size
		exclusion chromatography. 5
	(b)	What is capillary electrochromatography?
		Give its advantages over the parent
		technique. 5
	(c)	Explain the principle and the process of

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m SDS\text{-}PAGE}$ gel electrophoresis.