

**DIPLOMA - VIEP- MECHANICAL
ENGINEERING (DMEVI)**

Term-End Examination

June, 2016

00986

BIME-023 : ENGINEERING THERMODYNAMICS

Time : 2 hours

Maximum Marks : 70

Note : *Attempt five questions in all. Question no. 1 is compulsory. All questions carry equal marks. Scientific calculator is allowed. Assume missing data, if any. Use of steam table is permitted.*

1. Answer the following questions by choosing the best one out of the four given options : $7 \times 2 = 14$

(a) In S.I. units, the value of the universal gas constant is

(i) 0.8314 J/mol/K

(ii) 8.314 J/mol/K

(iii) 8314 J/mol/K

(iv) 83.14 J/mol/K

- (b) Kelvin Planck's law deals with
- (i) conservation of energy
 - (ii) conservation of heat
 - (iii) conservation of mass
 - (iv) conversion of heat into work
- (c) Entropy may be expressed as a function of
- (i) Pressure and Temperature
 - (ii) Temperature and Volume
 - (iii) Heat and Work
 - (iv) All of the above
- (d) The specific volume of water, when heated at 0°C
- (i) first increases then decreases
 - (ii) first decreases then increases
 - (iii) increases steadily
 - (iv) decreases steadily
- (e) For same compression ratio
- (i) thermal efficiency of Otto cycle is greater than that of diesel cycle.
 - (ii) thermal efficiency of Otto cycle is lesser than that of diesel cycle.
 - (iii) thermal efficiency of Otto cycle is equal to that of diesel cycle.
 - (iv) thermal efficiency of Otto cycle cannot be predicted.

(f) Petrol commercially available in India for Indian passenger cars has octane number in the range

- (i) 40 – 50
- (ii) 60 – 70
- (iii) 80 – 85
- (iv) 95 – 100

(g) Gibbs function is expressed as

- (i) $(U + pV - TS)$
- (ii) $(U + pV - TdS)$
- (iii) $(U + pdV - TdS)$
- (iv) $(U + pV - SdT)$

2. (a) What is thermodynamic system ? Differentiate between open system, closed system and isolated system, with examples. 8

(b) A tube contains an oil of specific gravity 0.9 to a depth of 120 cm. Find the gauge pressure at this depth in kN/m^2 . 6

3. (a) Explain the first law of thermodynamics, as referred to a closed system, undergoing a cyclic change. 6

(b) In an air motor cylinder, the compressed air has an internal energy of 450 kJ/kg at the beginning of the expansion and an internal energy of 220 kJ/kg after expansion. If the work done by the air during the expansion is 120 kJ/kg, calculate the heat flow to and from the cylinder. 8

4. (a) Give the following statements of the second law of thermodynamics : 6
- (i) Clausius Statement
 - (ii) Kelvin-Planck Statement
- (b) The coefficient of performance of a refrigerator operating on a reversed Carnot cycle is 5.6, when it maintains the temperature in the evaporator as -4°C . Determine the condenser temperature and refrigerating effect of the refrigerator, if the power required to run the refrigerator is 4.125 kW. 8
5. (a) Define the term availability. Also explain the concept of available and unavailable energy. What are the availability functions of flow and non-flow processes ? 6
- (b) How is the analysis of the products of combustion done ? Also discuss the methods to determine the composition of flue gases. 8
6. (a) Explain Mollier diagram in brief. Also discuss its use. 8
- (b) What amount of heat should be required to produce 4.4 kg of steam at a pressure of 6 kgf/cm^2 absolute and a temperature of 210°C from water at 30°C ? Take C_v for superheated steam as 0.5. 6

7. (a) What is air standard cycle ? Also discuss the assumptions made on air standard cycles.

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(b) Make a comparison between Carnot cycle and Rankine cycle. Also, drawing both the cycles on PV and TS diagram, discuss which one is more efficient.

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