

## P.G. DIPLOMA IN ANALYTICAL CHEMISTRY (PGDAC)

Term-End Examination

December, 2019

## MCH-004 : ELECTROANALYTICAL AND OTHER METHODS

Time : 3 hours

Maximum Marks : 75

*Note : (i) Attempt any five questions.**(ii) All questions carry equal marks.*

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1. (a) Write the Nernst equation for galvanic cell. Calculate the cell potential. 5  
(given  $a_{\text{Zn}^{2+}} = 5 \times 10^{-3}$ ,  $a_{\text{Cu}^{2+}} = 2 \times 10^{-2}$ )
  - (b) Write the classification of electroanalytical methods. 5
  - (c) How is the end-point detected in potentiometric titration? Explain with the help of plots. 5
  
  2. (a) Explain the alkaline error during pH measurement. 5
  - (b) Why does accuracy increase in pH titrations compared to direct pH measurements? 5
  - (c) How can glass electrodes be made selective for ions other than hydrogen ion? Give some examples. 5
  
  3. (a) Calculate the dissociation constant of  $0.05 \text{ mol dm}^{-3}$  ethanoic acid if its molar conductivity is  $16.3 \times 10^{-4} \text{ sm}^2 \text{ mol}^{-1}$  and limited conductivity is  $3.9 \times 10^{-2} \text{ sm}^2 \text{ mol}^{-1}$ . 5
  - (b) Beside concentration, which other factors affect the conductivity of the electrolyte solution? 5
  - (c) Explain the importance of over voltage in polarography. 5
  
  4. (a) A solution containing 0.75 g of copper as  $\text{Cu}^{2+}$  requires one hour for complete deposition of copper at 1.25 A. Calculate the coulombs required and efficiency of the process. ( $M_t = 63.54$ ) 5
  - (b) Describe the solubility product of a sparingly soluble salt with a suitable example. 5
  - (c) Discuss all the criteria for diagnosing reversible and irreversible redox reactions by cyclic voltammetry. 5
  
  5. (a) Differentiate between Linear Scan Polarography and Pulse Polarography. 5
  - (b) Define Diffusion Current with the help of Ilkovic's equation in polarography. 5
  - (c) Explain the significance of half wave potential in polarography. 5

6. (a) What is Biamperometry ? Draw the types of curves for detection of end point. 5  
(b) Discuss any five sources of errors in a thermogravimetric analysis. 5  
(c) Explain the principle of Differential Thermal Analysis (DTA). 5
7. (a) Explain Pulse Height Analysis (PHA) used for recording gamma ray spectra. Define Resolution with the help of spectrum. 5  
(b) What are radio tracer techniques ? Explain its advantages over conventional techniques. 5  
(c) Explain Radio Immuno Assay (RIA) and discuss its methodology. 5
8. Write brief notes on any five of the following : 5x3=15  
(a) Derivative Neutron Activation Analysis (DNAA)  
(b) Enthalpograms in Thermometric Titrations  
(c) Stripping Voltammetry  
(d) Limiting Molar Conductivity  
(e) Geiger Muller Counter  
(f) Solid State Membrane Electrodes
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