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BET-013

Diploma in Civil Engineering / Diploma in Electrical & Mechanical Engineering

01485

Term-End Examination June, 2010

BET-013: CHEMISTRY

Time: 2 hours Maximum Marks: 70 Note: Question number 1 is compulsory. Answer any other four questions from questions number 2 to 8. All questions carry equal marks. 1. What is Law of Octaves? (a) 2 2 Write the structure of Nitric Acid. Give two important ores of copper, with 2 their chemical formula. Why ice has lower density than water? 2 What is biogas and what are its uses? 2 (e) (f) What are sliding and rolling frictions? 2 What do you understand by Annealing? (g) What are Lanthanides and Actinides? Give 2. 6 two examples each. Write the electronic configuration of the following: Na⁺ (atomic no. of Na is 11) Cl⁻ (atomic no. of Cl is 17) Kr (36) Cs (55)

3.	(a)	Describe the Contact Process for commercial production of Sulphuric Acid.	6
	(b)	State any four uses of Hydrogen.	4
	(c)	What is Ozone and why ozone layer in atmosphere is important?	4
4.	(a)	Describe the functioning of Blast furnace with a neat diagram.	8
	(b)	Explain the three types of alloys with relevant examples.	6
5.	(a)	Discuss about scale formation and different conditioning methods.	8
	(b)	pH of a solution is 3. What is the pOH ? $(pK_w = 14)$	3
	(c)	Name any three units for expressing hardness of water.	3
6.	(a)	Explain any four criteria for selection of coal.	8
	(b)	Differentiate between water gas and producer gas.	2
	(c)	Define the calorific value of a fuel and mention any two units to express it.	4
7.	(a)	Define a lubricant and state any three functions of a lubricant.	6
	(b)	Differentiate between oil in water and water in oil emulsions.	4
	(c)	Write a note on Aniline point.	4

(a) What is PVC and write the chemical 4 formula of the monomer in it? (b) Write a detailed note on condensation 6 polymers. (c) Explain about vulcanisation.

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