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Ph.D. IN DAIRY SCIENCE AND TECHNOLOGY (PHDDR)

Term-End Examination

00252

June, 2015

RDR-011 : DAIRY AND FOOD ENGINEERING - I

Time : 3 hours

Maximum Marks : 100

Note : Attempt any **five** questions. All questions carry equal marks. Use of Steam Table is allowed (SI units).

- 1. What is the principle of spray drying ? Draw a neat sketch of a spray drier with various components. Compare the performance of pressure nozzle atomizer with centrifugal atomizer.
- 2. (a) Why is plate heat exchanger preferred over shell and tube heat exchanger in the dairy industry?
 - (b) What is the principle of working of a scraped surface heat exchanger ? Explain the construction features and performance characteristics.

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- (a) What engineering properties do you require for the design of an evaporator in the dairy industry ? Describe the method of evaluating any three properties.
 - (b) Explain hardness, adhesiveness, chewiness and gumminess.
- 4. (a) Classify various mixing equipments for mixing liquids. Explain where propeller, paddle and turbine type impellers are used in the dairy industry.
 - (b) How do you determine the power consumption of an agitator?
- 5. (a) Describe cone plate viscometer for determining the viscosity of non-newtonian fluid.
 - (b) Explain texture analyser used for measuring textural properties of any semi-solid food. 10
- 6. Write short notes on any *four* of the following: $4 \times 5 = 20$
 - (a) Climbing Film Evaporator
 - (b) Kelvin Model
 - (c) Fluidised Bed Drier
 - (d) Cyclone Separator
 - (e) Capillary Viscometer
 - (f) Plate Evaporator

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7. Milk of 13% solids is being concentrated in a single effect evaporator with a feed rate of 1500 kg/h at 20°C. The evaporator is being operated at sufficient vacuum to allow the product to evaporate at 70°C while steam is being supplied at 198.5 kPa. The desired concentration of the final product is 50% solids. Compute the steam requirement and steam economy for the process when the condensate is released at 70°C. Specific heat of 13% solid milk is 4.249 kJ/kg K and specific heat of water = 4.186 kJ/kg K.

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