

**Ph. D. (CHEMISTRY)/M. Phil.  
(CHEMISTRY)  
(PHDCHEM/MPHILCHEM)**

**Term-End Examination**

**June, 2022**

**RCHE-001 : ADVANCES IN ORGANIC CHEMISTRY**

*Time : 3 Hours*

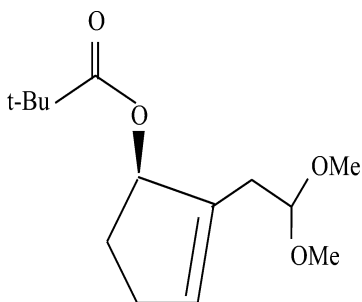
*Maximum Marks : 100*

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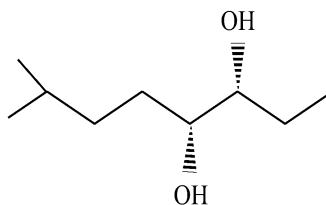
**Note :** Answer *all* questions.

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1. Suggest synthesis of a single enantiomer of any **one** of the following compounds : 10



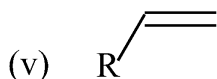
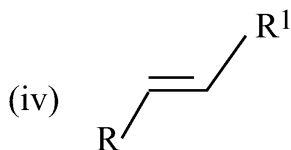
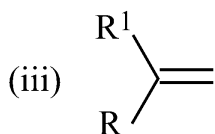
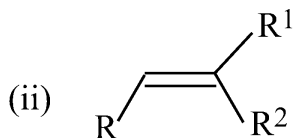
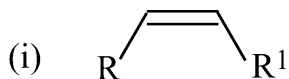
Or



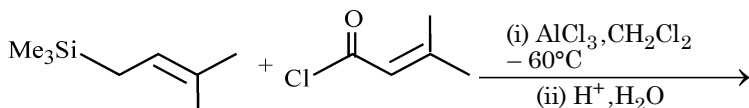
Write the name and structure of the reagent responsible for the asymmetric synthesis step.

**P. T. O.**

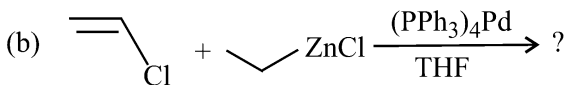
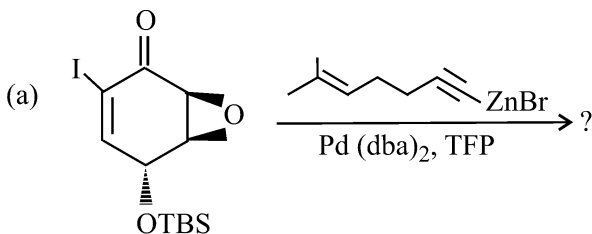
2. What are microwave assisted chemical reactions ? How do microwaves help in speeding up the reactions ? Explain with the help of an example. 10
3. What are the advantages of homogeneous catalysis over heterogeneous catalysis ? Arrange the following alkenes in the order of decreasing reactivity towards homogeneous catalytic hydrogenation : 10



4. Compare the stability of allylsilanes with lithium or boron allylic species towards organic synthesis. Complete the following reaction and give its mechanism : 10

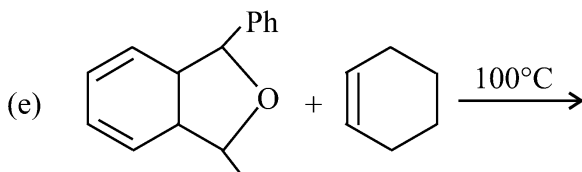
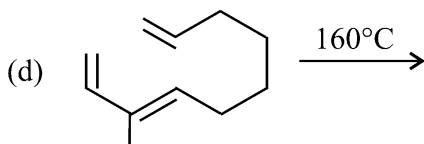
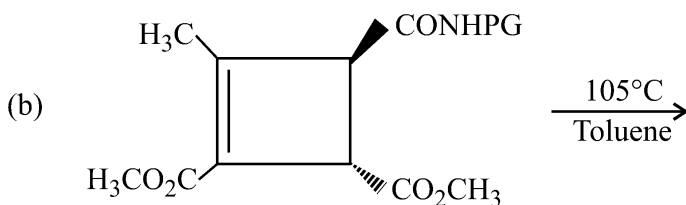
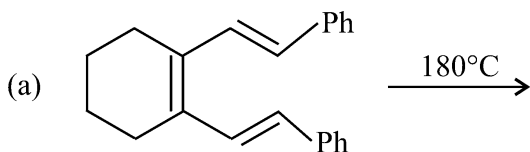


5. What are the challenges in carrying out solventless reactions especially in the context of organic synthesis ? Giving complete reaction conditions, write *two* reactions that can be performed solvent free. 10
6. Write the product in any *one* of the following complete reactions : 10



7. Draw the MO correlation diagrams for the electrocyclic reaction of 1, 3-butadiene for symmetry forbidden disrotatory and symmetry allowed conrotatory reactions. 10

8. Complete the following reactions :  $5 \times 2 = 10$



9. Taking suitable examples, discuss the forces that are responsible for supramolecular assemblies.

10. Complete the following reactions :  $5 \times 2 = 10$

