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MCH-003

**POST GRADUATE DIPLOMA IN
ANALYTICAL CHEMISTRY (PGDAC)**

Term-End Examination

June, 2021

MCH-003 : SPECTROSCOPIC METHODS

Time : 3 Hours

Maximum Marks : 75

Note : (i) Answer any **five** questions out of the eight given.

(ii) All questions carry equal marks.

1. (a) State Beer-Lambert's law. Write the factors responsible for the deviation of the law. 5
- (b) Draw a neat sketch of double beam spectrophotometer and explain the function of each component. 5

- (c) List various types of sources employed in IR spectrophotometry and explain any *one* in brief. 5
2. (a) How is FTIR different from IR ? Write any *two* advantages of FTIR spectrometer. 5
- (b) The molar absorptivity of a substance is $2.0 \times 10^4 \text{ cm}^{-1} \text{ mol}^{-1} \text{ dm}^3$. Calculate the transmittance through a cuvette of path length 5 cm containing $2.0 \times 10^{-6} \text{ mol dm}^{-3}$ solution of the substance. 5
- (c) Draw a schematic representation of the experimental setup of a Raman spectrometer and briefly explain the functions of its components. 5
3. (a) Describe in brief fluorescence quenching and write the Stern-Volmer equation. Which type of atoms/molecules show quenching ? 5

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- (b) Describe the analysis of amino acids and proteins by fluorimetry. 5
- (c) Discuss the factors affecting qualitative analysis by fluorimetry. 5
4. (a) State the principle of flame photometry. Write the important reactions occurring in the flame. 5
- (b) Describe the construction of a premix burner with a diagram. 5
- (c) Explain internal standard method used in flame photometric determination. 5
5. (a) Draw a neat diagram of Hollow Cathode Lamp (HCL) and explain its working. 5
- (b) Describe applications of AFS briefly. 5
- (c) Name any *five* types of atomic fluorescence transitions. 5
6. (a) Write a short note on atomisers used in AAS. 5
- (b) Write *three* advantages and *two* disadvantages of GFAAS. 5
- (c) Describe in brief interferences observed in AAS. 5

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7. (a) Write any *five* characteristics of an ideal atomization-excitation source used in ICP-AES. 5
- (b) Explain, how qualitative analysis is carried out using ICP-AES. Also state its importance. 5
- (c) Describe the salient features of AAS. 5
8. (a) Write the structure of TMS. Why is it used as a reference for chemical shifts in NMR ? 5
- (b) Discuss NMR spectrum of C₂H₅OH in low and high resolution. 5
- (c) Explain chemical ionization method to produce ions in mass spectrometer taking C₂H₅NH₂ as an example. 5

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