P.T.O.

BET-013

Diploma in Civil Engineering / Diploma in Electrical & Mechanical Engineering

DCLEVI/DMEVI/DELVI/DECVI/DCSVI/ACCLEVI/ACMEVI/ACELVI/ACECVI/ACCSVI

Term-End Examination June, 2012

BET-013: CHEMISTRY

| e: 2n | ours Maximum Marks | Maximum Marks : /0 | |
|-------|--|--|--|
| fo | our question from question numbers 2 to 8. All quest | | |
| (a) | What is Modern Periodic Table. | 2 | |
| (b) | Name the two allotropic forms of Oxygen. | 2 | |
| (c) | What is the difference between mineral and ore ? | 2 | |
| (d) | Why does ice float on water? | 2 | |
| (e) | Enlist any four gaseous fuels. | 2 | |
| (f) | What is a copolymer? Give an example. | 2 | |
| (g) | Write any four uses of bleaching powder. | 2 | |
| (a) | Why an anion is always larger than its parent atom? | 4 | |
| (b) | Explain the following. | 6 | |
| | (i) Dobereiner triad | | |
| | (ii) Law of octaves | | |
| | (iii) Inner transition elements | | |
| | (a) (b) (c) (d) (e) (f) (g) (a) | e: Question number 1 is compulsory. Answer any o four question from question numbers 2 to 8. All questicarry equal marks. (a) What is Modern Periodic Table. (b) Name the two allotropic forms of Oxygen. (c) What is the difference between mineral and ore? (d) Why does ice float on water? (e) Enlist any four gaseous fuels. (f) What is a copolymer? Give an example. (g) Write any four uses of bleaching powder. (a) Why an anion is always larger than its parent atom? (b) Explain the following. (i) Dobereiner triad (ii) Law of octaves | |

1

| (| (c) | Write the electronic configuration of : | 4 |
|----|---|---|---|
| | | (i) He (2) | |
| | | (ii) Ne (10) | |
| | | (iii) Cl (17) | |
| | (iv) Mg ²⁺ (Atomic number of Mg is 12) | | |
| 3. | (a) | Write short notes on the following: | , |
| | | (i) Atmosphere | 3 |
| | | (ii) Important zones of atmosphere | 3 |
| | | (iii) Biogas | 2 |
| | (b) | Write the chemical formula of following compounds | 4 |
| | | (i) Hydrochloric Acid | |
| | | (ii) Sulphuric Acid | |
| | | (iii) Nitric Acid | |
| | | (iv) Nitrogen dioxide. | |
| | (c) | Write the isotopes of hydrogen. | 2 |
| 4. | (a) | Explain the function of a Blast furnace with a neat diagram. | 8 |
| | (b) | How is copper purified? Explain. | 3 |
| | (c) | What is high carbon steel? Give its important uses. | 3 |
| 5. | (a) | Explain the hydrological cycle, with a diagram | 6 |
| | (b) | What is hydrogen bond? Explain with a diagram. | 4 |
| | (c) | pH of a solution is 3. What is the pOH of the solution ? (pKw=14) | 4 |

6. (a) Explain the construction of a Bomb 8 Calorimeter with a neat diagram. What are primary and secondary fuels? 6 (b) Give 3 examples for each type. Write a short note on following types of 7. (a) 8 lubricants. Liquid lubricants (i) Semi - solid lubricants (ii) (iii) Solid lubricants Synthetic lubricants (iv) (b) Distinguish between Flash point and fire 4 point. Write about sliding friction. 2 (c) 8. (a) Differentiate between isotactic, syndiotactic, 6 and atactic polymers on the basis of their structure. Write a detailed note on composition, (b) 6 properties and uses of any two types of glasses. (c) Why glass is called as "super cooled liquid"? 2