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MEV-013

M. SC. (ENVIRONMENTAL SCIENCE) (MSCENV) Term-End Examination December, 2022 MEV-013 : ENVIRONMENTAL CHEMISTRY

Time : 3 Hours			Maximum Marks : 100		
Note : Answer	any	ten	questions.	All	questions
carry equ	ual m	arks.			

- What are the principles of Green Chemistry ?
 Explain in detail.
 10
- Describe the mechanism of ozone formation in the lower troposphere.
 10
- 3. Explain the reasons for the following : 5+5
 - (a) Aluminium sulphate is used as a coagulant
 - (b) Ozone hole not seen in non-polar regions
- 4. Differentiate between the primary and secondary particulate matter with examples. 10

- 5. Explain the toxic effects of the following on humans: 5+5
 - (a) Cyanide
 - (b) Ozone
- 6. Explain the carbonate system in oceans. 10
- 7. Write short notes on the following : 5+5
 - (a) Radionuclides
 - (b) Tetra Ethyl Lead (TEL)
- 8. Explain the mechanism of adsorption. 10
- What are UV-A, UV-B and UV-C radiation ?
 Explain the significance of UV-B radiation. 10
- 10. Define the following : $2\frac{1}{2} \times 4 = 10$
 - (a) Polarography
 - (b) Chemical potential
 - (c) Silo-filler disease
 - (d) Pedogenesis
- 11. Explain basic instrumentation of Mass spectrometry. What are its environmental applications?

- 12. What is the working principle of High-Performance Liquid Chromatography ? Describe its applications.
- 13. Write short notes on any *two* of the following :

5 + 5

- (a) Acid mine drainage
- (b) K-type and F-type wastes
- (c) Photochemical Smog
- 14. Explain the relation between atmospheric stability and vertical mixing.10

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