No. of Printed Pages: 4

**MCH-002** 

## P.G. DIPLOMA IN ANALYTICAL CHEMISTRY (PGDAC)

## **Term-End Examination**

00062

December, 2018

## MCH-002 : SEPARATION METHODS

Time : 3 hours

Maximum Marks: 75

**Note :** Attempt any **five** questions. All questions carry equal marks.

1. (a) Explain the evolution of chromatography. Describe the principle involved in the chromatographic separations.

(b) Name the separation methods based on

- (i) volatility,
- (ii) solubility, and
- (iii) ion exchange phenomena.

Give principle of any one of them.

(c) Discuss the principle of paper chromatography. 5

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- (a) Explain multiple extractions with successive equal portions of the extractant and derive expression for weight of solute after n extractions.
  - (b) Briefly discuss extraction by solvation. Explain the formation of  $FeCl_4$  anion in the aqueous phase during extraction of Fe (III) by this process.
  - (c) What is a diluent ? Give reasons for its use during solvent extraction.
- 3. (a) Explain (i) retention time  $(t_R)$ , and (ii) retention factor  $(k'_X)$ . How does too small  $k'_X$  value affect the elution ?
  - (b) Explain the concept of theoretical plates in column chromatography. Calculate the number of theoretical plates if retention time  $(t_R) = 4.37$  min and half peak width  $(W_{1/2}) = 0.63$  min.
  - (c) Explain the term resolution of a chromatogram with the help of a suitable illustration. Calculate resolution if  $t_x = 5.83 \text{ min}, t_y = 7.25 \text{ min}, W_{x/2} = 0.57 \text{ min}$  and  $W_{y/2} = 0.71 \text{ min}.$

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- 4. (a) What are the various methods of chromatographic development ? Explain any one of these methods with an example.
  - (b) Explain the methodology of sample application and detection in HPTLC.
  - (c) Explain the term solvent efficiency or column efficiency with the help of suitable illustration and give its mathematical expression.
- 5. (a) What are the essential requirements of a good detector in gas chromatography?
  - (b) Explain the principle of high performance liquid chromatography (HPLC).
  - (c) Explain basic principle and operation of gel electrophoresis.
- 6. (a) Discuss unique features of size exclusion chromatography.
  - (b) Explain thermospray method of interfacing of HPLC with mass spectrometry.
  - (c) What are the various resin properties ?Explain any one of these. 5

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- 7. (a) Define gels and describe their important properties.
  - (b) What are synthetic inorganic ion exchangers ? Give their different types.
  - (c) Briefly explain capillary electrophoresis giving a schematic diagram.
- 8. Write notes on any *three* of the following with suitable illustrations or examples wherever possible:  $3 \times 5 = 15$ 
  - (a) Desalination and Water Treatment
  - (b) Ion Selective Membrane Electrode
  - (c) Electrodialysis
  - (d) Membrane Separation Process

1,500

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