No. of Printed Pages: 4

**RCHE-002** 

## Ph.D. IN CHEMISTRY (PHDCHEM)

00431

## **Term-End Examination**

## December, 2017

## RCHE-002: ADVANCES IN INORGANIC **CHEMISTRY**

Time: 3 hours

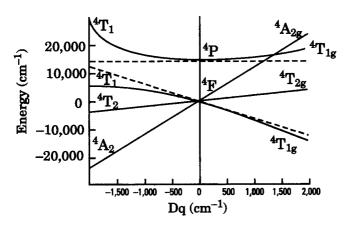
Maximum Marks: 100

**Note:** Answer all the questions.

Illustrate the Jahn-Teller effect, with suitable 1. 10 examples.

(a) What information do you get from the 2. following Orgel diagram?

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- (b) Give the configurations of a d<sup>4</sup> metal ion in both, a weak field and a strong field, and calculate the respective CFSE in an octahedral field.
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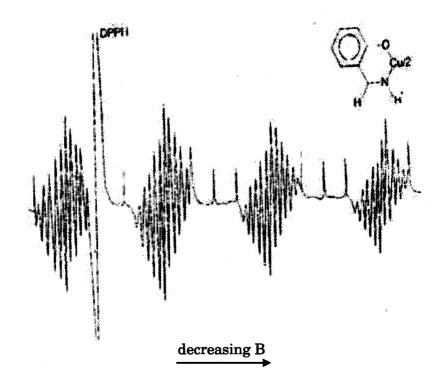
- 3. (a) Calculate the spin-only moment and spin-plus orbital moment of manganese(II) ion.
  - (b) "In general, orbital contribution is more likely in square planar geometry than in octahedral geometry." Critically evaluate the statement, giving reasons.
  - (c) What will the Russell-Saunders symbol of Ni(II) ion be? Write the steps involved. 10
- 4. (a) Draw a rough sketch of the expected <sup>31</sup>P{<sup>1</sup>H} and <sup>109</sup>Rh NMR spectra of the tetrahedral complex, [Rh(NO)(PC<sub>6</sub>H<sub>5</sub>)<sub>3</sub>]<sub>3</sub>
  - (b) Given that:

 $^{10}\text{B}: Q = 8.5^*\ 10^{-2}, \ a = 19.6\%, \ r^N = 10.7, \ I = 3$   $^{11}\text{B}: Q = 4.1^*\ 10^{-2}, \ a = 80.4\%, \ r^N = 32.1, \ I = \frac{3}{2},$  draw the  $^{19}\text{F}$  NMR spectrum of NaNBF<sub>4</sub> molecule in D<sub>2</sub>O, assuming that there is a coupling between the two isotopes of boron with fluorine (I =  $\frac{1}{2}$  and 100% abundance). Justify your answer.

*10* 

(c) Explain the following first derivative ESR spectrum of bis (salicyaldiminato) copper(II) with isotopically pure <sup>63</sup>Cu.

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- 5. (a) Outline the similarities and differences in the basic structure of heme in hemoglobin and chlorophyll.
  - (b) What is the Prosthetic Group in Cytochromes?
    Write its physiological role.
  - (c) What makes Fe and Cu suitable for redox processes in biological systems?

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P.T.O.

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6.	(a)	"Much of the inspiration and origins of	
		supramolecular chemistry comes from the	
		chemistry found in living biological	
		systems." Comment on this statement.	5
	(b)	What are Zwitterions? In biological systems,	
		how are they found?	5
	(c)	How are the Edge-directed and Face-directed	
		self-assembling coordination compounds	
		synthesised?	5

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