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BET-013

DIPLOMA IN CIVIL ENGINEERING DCLE(G) / DIPLOMA IN MECHANICAL ENGINEERING (DME) / DCLEVI / DMEVI / DELVI / DECVI / DCSVI / ACCLEVI / ACMEVI / ACELVI / ACECVI / ACCSVI

77200

Term-End Examination

December, 2017

BET-013 : CHEMISTRY

Time : 2 hours

Maximum Marks: 70

- **Note:** Question number 1 is compulsory. All questions carry equal marks. Attempt any four questions from questions no. 2 to 8.
- What is the Aufbau Principle? 1. **(a)**
 - What is Modern Periodic Law? **(b)**
 - (c) What is Polymerization Process?
 - (d) Define Glass Transition Temperature (T_{σ}) .
 - Define Triad with suitable examples. (e)
 - **(f)** Complete the following reaction :

 $CaCl_2 + ... \rightarrow CaCO_3 + 2NaCl$

Ionic compounds do not conduct electricity (g) in a solid state. Explain. $7 \times 2 = 14$

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2.	(a)	Describe the disadvantages of hard water in a boiler.	8
	(b)	Why does ice float on liquid water ?	4
	(c)	Write the advantages of using chlorine as a disinfectant.	2
3.	(a)	Define Ionization Energy (IE). What are the factors which affect IE ?	10
	(b)	IE of nitrogen is greater than oxygen. Explain why.	2
	(c)	State the causes of Periodicity.	2
4.	(a)	Describe the procedure for determining the calorfic value of a fuel by bomb calorimeter.	8
	(b)	Classify Lubricants. Give two examples of solid lubricants.	3
	(c)	What is Flash Point ? How is it different from Pour Point ?	3
5.	(a)	Name any two ores of Aluminium.	2
	(b)	What is Red Mud ? Explain.	4
	(c)	Explain the process of extraction of aluminium.	8

6.	(a)	Name the monomer of natural rubber and	
		explain its vulcanization process. 10	I
	(b)	What is the role of plasticizers ? 2	
	(c)	Differentiate between the properties of low	
		density and high density polyethylene. 2	
7.	(a)	Describe the physical and chemical	
		properties of bleaching powder. 10	I
	(b)	Explain why silicates, phosphates and	
		borates have a greater tendency to form	
		glass. 2	
	(c)	How are refractory materials classified? 2	
8.	Write short notes on any <i>two</i> of the following : $2 \times 7 = 14$		
	(a)	Functions of Borax and Alumina in	
		Manufacturing of Glass	
	(b)	Properties of Sulphuric Acid	
	(c)	Functions of Lubricants	

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